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#### Key indicators

Single-crystal X-ray study  
 $T = 293\text{ K}$   
 Mean  $\sigma(\text{C}-\text{C}) = 0.003\text{ \AA}$   
 $R$  factor = 0.037  
 $wR$  factor = 0.109  
 Data-to-parameter ratio = 14.1

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

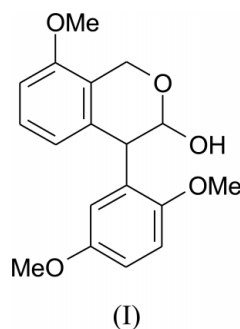
## (3*S*,4*R*)-4-(2,5-Dimethoxyphenyl)-8-methoxyisochroman-3-ol

In the title compound,  $\text{C}_{18}\text{H}_{20}\text{O}_5$ , the hydroxyl and dimethoxyphenyl substituents are in axial positions. The heterocyclic ring is in a half-chair conformation. The molecules are linked by  $\text{O}-\text{H}\cdots\text{O}$  hydrogen bonds, leading to the formation of a chain extended throughout the whole of the crystal.

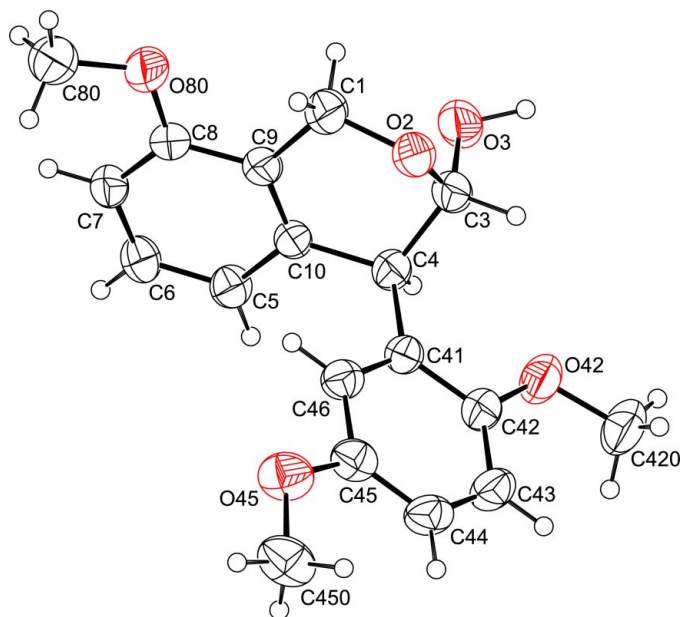
Received 15 April 2002  
 Accepted 23 April 2002  
 Online 11 May 2002

#### Comment

Some members of the benzo[*c*]pyran family have been found in nature and have been shown to possess a variety of biological properties (Moore, 1977; Moore & Czerniak, 1981). The title isochroman, (I), has been synthesized as part of a research project on new precursors for obtaining this type of antibiotic (Epszajn *et al.*, 2001). X-ray investigations were made in order to confirm the configurations of the chiral atoms C3 (*S*) and C4 (*R*), and to define the positions of the substituents attached to these atoms.



The molecule of the title compound consists of two condensed rings, phenyl and heterocyclic, with atom O2 in position 2 of the latter. There is a hydroxyl group in position 3, a 2,5-dimethoxyphenyl substituent in position 4 and a methoxy group in position 8. The heterocyclic ring has a half-chair conformation, with the twofold axis passing through the midpoint of the O2–C3 bond. The puckering parameters (Cremer & Pople, 1975) corresponding to the sequence C1–O2–C3–C4–C10–C9 are  $Q = 0.477$  (2)  $\text{\AA}$ ,  $\varphi_2 = -77.6$  (4) $^\circ$  and  $q_2 = 132.6$  (3) $^\circ$ , and the asymmetry parameter (Nardelli, 1983)  $\Delta_2(\text{O2}-\text{C3})$  is 0.0413 (8). The substituents in positions 3 and 4 of the heterocyclic ring are attached axially, with torsion angles O3–C3–C4–C10 and C41–C4–C10–C9 of  $-70.7$  (2) and  $101.8$  (2) $^\circ$ , respectively. The phenyl rings are almost planar and form a dihedral angle of  $85.09$  (6) $^\circ$ . The O atom of the hydroxyl group acts as a hydrogen-bond donor to O2<sup>i</sup> of an adjacent molecule [symmetry code: (i)  $-x+5/2, -y, z+1/2$ ; see Table 2]. Finally, the linked molecules form a *C*(4) chain (Fig. 2) (Bernstein *et al.*, 1995).



**Figure 1**  
Displacement ellipsoid plot (*PLATON*; Spek, 1998) of title compound, with the atom-labelling scheme. Ellipsoids are drawn at the 40% probability level.

## Experimental

The synthesis of the title compound has been described elsewhere (Epsztajn *et al.*, 2001). Crystals were obtained by slow evaporation from a methanol solution at room temperature.

### Crystal data

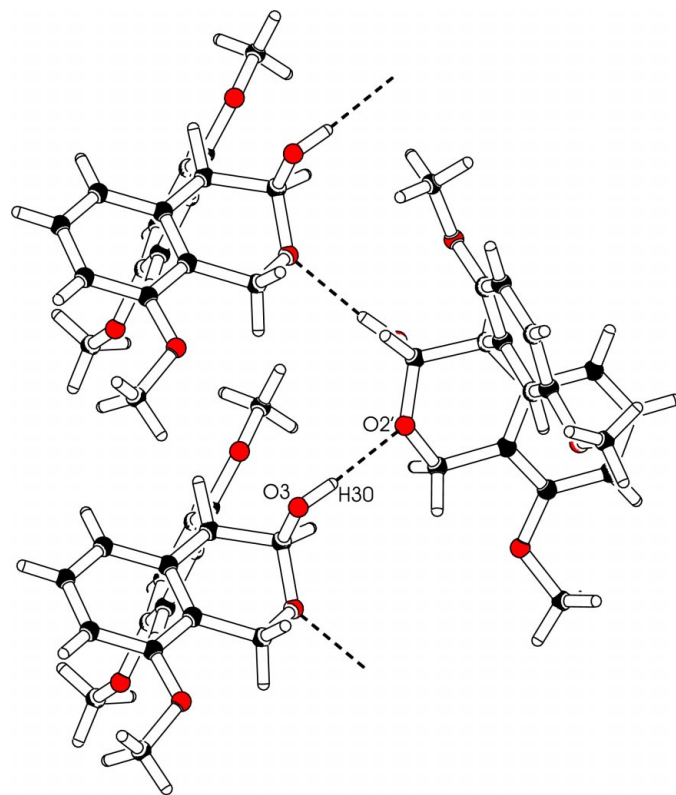
$C_{18}H_{20}O_5$	Cu $K\alpha$ radiation
$M_r = 316.34$	Cell parameters from 20 reflections
Orthorhombic, $P2_12_12_1$	$\theta = 9.5\text{--}12.3^\circ$
$a = 11.920$ (1) Å	$\mu = 0.79$ mm $^{-1}$
$b = 18.114$ (2) Å	$T = 293$ (2) K
$c = 7.383$ (1) Å	Needle, colourless
$V = 1594.2$ (3) Å $^3$	$0.50 \times 0.28 \times 0.18$ mm
$Z = 4$	
$D_x = 1.318$ Mg m $^{-3}$	

### Data collection

Rigaku AFC-5S diffractometer	$R_{\text{int}} = 0.024$
$\omega$ scans	$\theta_{\text{max}} = 72.6^\circ$
Absorption correction: analytical (de Meulenaer & Tompa, 1965)	$h = -14 \rightarrow 14$
$T_{\text{min}} = 0.778$ , $T_{\text{max}} = 0.879$	$k = -22 \rightarrow 22$
6547 measured reflections	$l = -8 \rightarrow 8$
3063 independent reflections	3 standard reflections every 150 reflections
2599 reflections with $I > 2\sigma(I)$	intensity decay: <2%

### Refinement

Refinement on $F^2$	$w = 1/[\sigma^2(F_o^2) + (0.0726P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.037$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.109$	$(\Delta/\sigma)_{\text{max}} < 0.001$
$S = 1.02$	$\Delta\rho_{\text{max}} = 0.17$ e Å $^{-3}$
3063 reflections	$\Delta\rho_{\text{min}} = -0.15$ e Å $^{-3}$
217 parameters	Extinction correction: <i>SHELXL97</i>
H atoms treated by a mixture of independent and constrained refinement	Extinction coefficient: 0.0125 (9)
	Absolute structure: Flack (1983), 1281 Friedel pairs
	Flack parameter = $-0.1$ (2)



**Figure 2**  
The intermolecular hydrogen bonding in the crystal structure of the title compound.

**Table 1**

Selected geometric parameters (Å, °).

O2—C3	1.426 (3)	O45—C450	1.416 (2)
O2—C1	1.436 (2)	C42—O42	1.367 (2)
O80—C8	1.369 (2)	O3—C3	1.406 (2)
O80—C80	1.420 (3)	O42—C420	1.416 (3)
O45—C45	1.383 (3)		
C3—O2—C1	113.8 (2)	O3—C3—O2	110.7 (2)
C8—O80—C80	117.9 (2)	O3—C3—C4	110.0 (2)
C45—O45—C450	118.2 (2)	O2—C3—C4	109.8 (2)
C42—O42—C420	118.8 (2)		
C9—C10—C4—C41	101.8 (2)	O3—C3—C4—C10	$-70.7$ (2)

**Table 2**

Hydrogen-bonding geometry (Å, °).

$D\text{—}H\cdots A$	$D\text{—}H$	$H\cdots A$	$D\cdots A$	$D\text{—}H\cdots A$
O3—H30 $\cdots$ O2 $^i$	0.98 (4)	1.90 (4)	2.881 (2)	171 (3)

Symmetry code: (i)  $\frac{1}{2} - x, -y, \frac{1}{2} + z$ .

Atoms H30 and H3 were refined isotropically. The other H atoms were constrained to ride on their parent C atoms using AFIX in *SHELXL97* (Sheldrick, 1997).

Data collection: *MSC/AFC Diffractometer Control Software* (Molecular Structure Corporation, 1989); cell refinement: *MSC/AFC Diffractometer Control Software*; data reduction: *TEXSAN* (Molecular Structure Corporation, 1989); program(s) used to solve structure:

*SHELXS86* (Sheldrick, 1990); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *PLATON* (Spek, 1998); software used to prepare material for publication: *PARST97* (Nardelli, 1996).

Financial support from the University of Łódź (grant No. 505/667) is gratefully acknowledged.

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